University of Calgary Department of Chemical & Petroleum Engineering

ENCH 501: Transport Processes

Mid-Term Examination, Fall 2007

Instructions: Time: 2:00 to 3:30 pm Oct 23, 2007

Attempt All Questions. Open Notes & Book.

Use of calculators permitted

Problem #1 (15 points)

Inadequate sterilization of fluids or components of medical devices and surgical instruments, and the significant adverse health risks to patients, have been in the news recently. Pathogenic microorganisms need to be destroyed or removed from contact lenses (e.g. *Acanthamoeba*) or from forceps, scapels, components of dialysis machines, clothes and many other surfaces which come in contact with blood in hospitals. The use of both chemical disinfectants (such as bleach and hydrogen peroxide) and heat are encouraged for sterilization. It is also desirable to remove biofilms (contaminated with viruses, mold and bacteria) which might have coated the surfaces. The problem of current interest involves heat sterilization in an autoclave which functions like a pressure cooker. The object to be sterilized has to be raised to a temperature of 121°C and held at or above this temperature for 60 minutes. Temperatures above 100°C are required to destroy *Bacillus* and *Clostridium* spores.

An autoclave for sterilizing steel surgical instruments is to be modelled. The capacity of the autoclave is 30 litres and it is heated through a steam jacket. Saturated steam at 2 atm (abs) and 123.3°C is passed into the jacket at controlled rates. Forty five (45) solid steel cylinders, each 5 cm diameter and 12.5 cm long, are placed in the autoclave which was then filled to capacity with water. At the start, both the cylinders and the water were at 16°C. Then steam was admitted into the jacket in a manner that the temperature of the water in the autoclave (T°C), recorded using a thermocouple, satisfies the following function:

$$T = 16 + \Delta T (1 - e^{-bt})$$

where ΔT is the maximum possible temperature rise for the water in the autoclave (123.3 - 16)°C, t is time in minutes and the exponent **b** equals 0.2135.

Given the data below and, assuming that the cylinders are stacked in such a way that each is substantially and freely exposed to the water which is well-mixed, the coefficient of convective heat transfer around each cylinder is 320 W/m² K,

- a) how long will be required to sterilize the cylinders?
- b) After how long would the temperature of the water in the autoclave reach 121°C? Comment on the results from parts (a) and (b).
- c) If the autoclave is insulated, how much total condensate, at saturated condition, will have been produced from the steam jacket at the instant the cylinders are sterilized, from the start of the process?

State and justify your assumptions. Show all your steps.

Data:

Properties of steel: k = 43 W/mK, $\rho = 7,801 \text{ kg/m}^3$, Cp = 0.473 kJ/kg K

Properties of water (avg): k = 0.654 W/mK, $\rho = 980$ kg/m³, Cp = 4.179 kJ/kg K, Latent heat of

vaporization at 123.3°C = 2,193 kJ/kg



Problem #2 (10 points)

A billboard, 1.5 m high and 2.5 m long, is attached to a wall of a large building. The board is loosely attached to the wall by brackets. You are concerned that a wind flowing parallel to the building at or above 10 km per hour may be strong enough to dislodge the billboard. You may assume that the air is flowing only over the exposed surface of the board in a direction parallel to the longer side.

What force must the brackets be able to withstand under these conditions? Use the *integral method* and assume that the velocity profile along the board satisfies

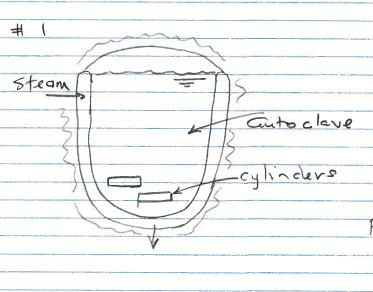
$$u = a + c \sin(by)$$

where a, b and c are constants or functions of x (distance from the leading edge along the direction of flow), and y is the direction normal to the board.

Show the important steps in your derivations.

Data:

Properties of air; μ = 0.01846 mPa s; ρ = 1.1774 kg/m³; k = 0.02624 W/m K; air temperature = 27°C.



The focus of the problem are the cylinders.

First check whether the Lumped Andysis Method

For a cylinder

h (/A) should be < 0.

 $\pi dh + 2\pi d^2 = \pi(5)(15^2)(12.5)(15^2)$ $= 2.3562(15^{-2})^{2} \frac{71(25)(15^{-4})}{7}$

 $\frac{320(2.4544)}{2.3562}(10) = 0.0775 < 0.1$

assume no temp. gradients in cylinders.

Total ustume occupated by cylinders 45 (+) = 0.011045 m3

since total volume = 30 litres 12 0.03 m3

the volume of water = 0.018955 m3

its mass = 980 (0.018955) = 18.5761 kg

With respect to each cylinder, the autoclave water is the ambient at To where Ta = 16 + (123.3-16)(1-e) $T_{a}(t) = 123.3 - 107.3 e^{-0.2135t} = temp. st$ $= a + ce^{-bt} = autoclav.$ Use a cylinder as the control where, every belance is: Input = Accum or hA(Ta-T) = map di or on substitution to Ta hA(a+cl-bt-T) = mCpdT Let B = hA/mCp :. B(a+ce-b+)-BT=dT dT + BT = P(t) = B(a+ce-bt) Use integration factor espect to obtain e Spot = (spot Q(t) at + Co = (lBt (Ba + Bcl-bt) oft + Co Test = alt + Be est - bt + Co T = a + BC 0-bt + Col → temperature

B-b cfmidle

with the unknown constant a determined

using the condition - t=0, T=16°C

```
\beta = hA \qquad 320(2.3562)(10^{-2})
m^{C}p \qquad 2.4544(10^{-4})(7801)(473)
              = 8,3254 (15-3) 5- or 0.4995 min
          = 123, 3 °C
         = -107.3°C
   Substitute, Co = +80.1088
  Hence, temperature of cylinder (°C)
T(t) = 123.3 - 187.409 e + 180.10882
                    7,0
        t, mui
                      16
        18
                     119.29 -> 123.3 - 4.014+0.009976
        25
                     122.4
                     121.59
                     121.18
                     120.95-123.3- 2.355 + 0.00286
        20.6
                     120,997->123.3-2.3052+01002722
  is total time required to sterilize cylinders
        = 20.6 + 60 = 80.6
            I hour 20 minutes and 365.
b) from
          Ta = 123.3 - 107.3 e
     When Tx = 121°C, t= 18 mis.
   Comment! We note that the water withe
```

autoclave attains 121°C feeter than

the cylinders. If it had been assumed that
the water and the cylinders will always
be at the same temperature, one would have
specified 78 minutes for the operation.

It is obvious, under given conclitions, that
some of the microsyvanisms may survive!

© At t = 80.6 minutes

-0.2135 (80.6)

water temp, $7x_f = 123.3 - 107.32$ = 123.3 °C

cylinder temp, Tp = 123.3 °C

The mass of condensate, m, equals

m = mwqw(Tef-16) + mCp(Tf-16)

= 18.5761 (4179) (123.3-16)+

45 (2.4544)(10-4)(7801)(473)(123.3-16)

(2193)(103)

= 8,329,647.7 + 4,372,894.47 $\frac{193(10^3)}{2193(10^3)}$

= 5.7923 k=

#2 Given U = 10 lem/hr = 2.778 m/s Check whether entre b.1. The b. 1. becomes two leat N=L when Re = U2 xp = 5(105) n=0 5(105) = (2.778) x (1-1774) 1.846 (10-5) x = 2.822 m. The vi longer than the side of the billboard. Itemse the b.l. will be laminar. The momentum integral equation is (Eq. 5.13, Notes) May = d [Sop(ua-u)udy] with the conditions y=0 u=0, no y= 5 n= 4 profile -Given = a + csin by duldy = bc cosby dzu/dz = - b2c sin by -> satisfies b.c (4)

Solve
$$S^2 = 22.9952 \frac{VX}{V_{0}}$$

Solve $S^2 = 22.9952 \frac{VX}{V_{0}}$
 $S = 4.7953 \frac{1}{2} \frac{1}{2}$

Drag on billboard, $D = \int_{C} f_{0} W dn = \frac{1}{2} \frac{1}{2}$
 $= W \int_{0}^{1} \frac{1}{4.7953} \frac{1}{2} \frac{1}{2} \frac{1}{2} \frac{1}{2} \frac{1}{2} \frac{1}{2}$
 $= W \int_{0}^{1} \frac{1}{4.7953} \frac{1}{2} \frac$